

K. Samikov, R. Shakirov,
and S. Yu. Yunusov

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Continuing an investigation of the epigeal part of *Korolkowia sewerzowii* Regel., the methanolic mother solution from korseveriline and the acetone mother solution from korseveridine [1] were chromatographed separately on a column of alumina. Elution was carried out with chloroform-methanol (10:0.3 and 10:0.5, respectively). The chloroform-methanol (10:0.3) eluate yielded a base with mp 182-183°C (benzene) (I), $[\alpha]_D -17^\circ$ (c 0.71; chloroform). The IR spectrum of (I) contained absorption bands at (cm^{-1}) 3400 (OH), 2950-2865 and 1440 ($-\text{CH}_3$, $-\text{CH}_2-$), and 1730 and 1260 (C=O of an ester group). Elution with chloroform-methanol (10:0.5) gave a base with mp 150-152°C (acetone) (II), $[\alpha]_D -13^\circ$ (c 0.7; chloroform), M^+ 427.

In addition to korsemine, the mother liquors from sevcorine [1, 2], on elution with benzene-methanol (10:0.5 and 10:1), yielded a base (III) with mp 205-207°C (acetone). The IR spectrum of (III) contained absorption bands at (cm^{-1}) 3430 (OH); 1740, 1250 (C=O of an ester group); and 1125-1030 (broad absorption band characteristic for glycoalkaloids).

When (II) was acetylated with acetic anhydride in pyridine, a O,N-diacetyl derivative (IV) was obtained with M^+ 511. Its IR spectrum contained absorption bands at (cm^{-1}) 1730, 1245 (O-Ac); 1710 (C=O); and 1665 (N-Ac). Its PMR spectrum contained singlets at (ppm) 0.62 (3 H, 19- CH_3); 0.72 (3 H, 18- CH_3); 1.96 (3 H, OCOCH_3); and 1.98 (3 H, N- COCH_3), and two doublets at 1.08 and 1.15 from secondary methyl groups.

A comparison of the physical constant and spectral characteristics of alkaloid (II) and of an authentic sample of petsidine [3-7] showed their identity (melting point of a mixture, R_f values, and IR, PMR, and mass spectra).

This is the first time that petsidine has been detected in plants from the genus *Korolkowia*.

LITERATURE CITED

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